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Key indicators

Single-crystal X-ray study
 $T = 298\text{ K}$
Mean $\sigma(\text{C}-\text{C}) = 0.005\text{ \AA}$
 R factor = 0.036
 wR factor = 0.068
Data-to-parameter ratio = 17.0For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.

1,6-Bis(phenylsulfanyl)hexane

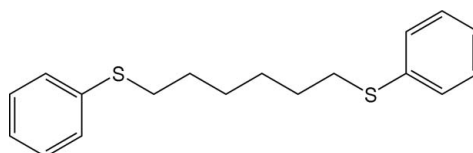
The title compound, $\text{C}_{16}\text{H}_{18}\text{S}_2$ (or L^6), crystallizes with two half-molecules in the asymmetric unit and each independent molecule lies about a crystallographic center of symmetry. The aliphatic segment of this ligand is in an all-*trans* conformation.

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Comment

Linear bifunctional ligands are usually used as building blocks for the construction of metal-organic framework materials (MOF) (Carlucci *et al.*, 2002). Sulfur, being a soft base, has good ability to coordinate to an Ag atom which is a soft acid. As the dithiolate ligand affords two coordination sites to metal centers, supramolecular architectures may be achieved (Black *et al.*, 1995; Bu *et al.*, 2002). This flexible ligand was synthesized in order to help understand the formation of supramolecular networks, which is affected by many factors, such as the type of solvents, the counter-anions, the metal-to-ligand ratio, the metal coordination and the ligand flexibility [see, for example, Withersby *et al.* (1997, 1999), Noro *et al.* (2002), Blake *et al.* (1999)]. We have recently determined the structures of the analogous compounds L^{10} and L^2 (Awaleh *et al.*, 2005*a,b*). We report here the structural characterization of 1,6-bis(phenylsulfanyl)hexane (L^6).

(I)

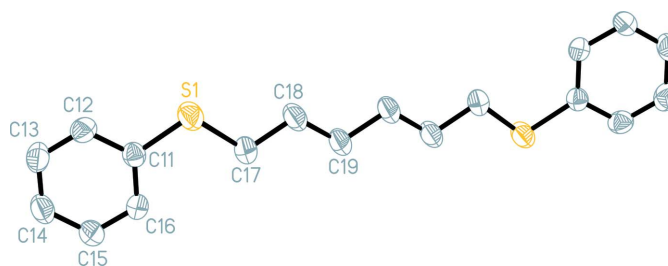
 L^6 crystallizes with two half-molecules in the asymmetric unit. A view of one of the molecules is shown in Fig. 1. Each of the two molecules has a center of symmetry at the mid-point of the central C—C bond, *viz.* C19—C19ⁱ [symmetry code: (i)

Figure 1

View of one molecule of L^6 , showing the atom-numbering scheme. Probability displacement ellipsoids are shown at the 50% level. The H atoms have been omitted. The unlabeled part of the molecule is related by the symmetry code $(-x, -y + 1, -z)$.

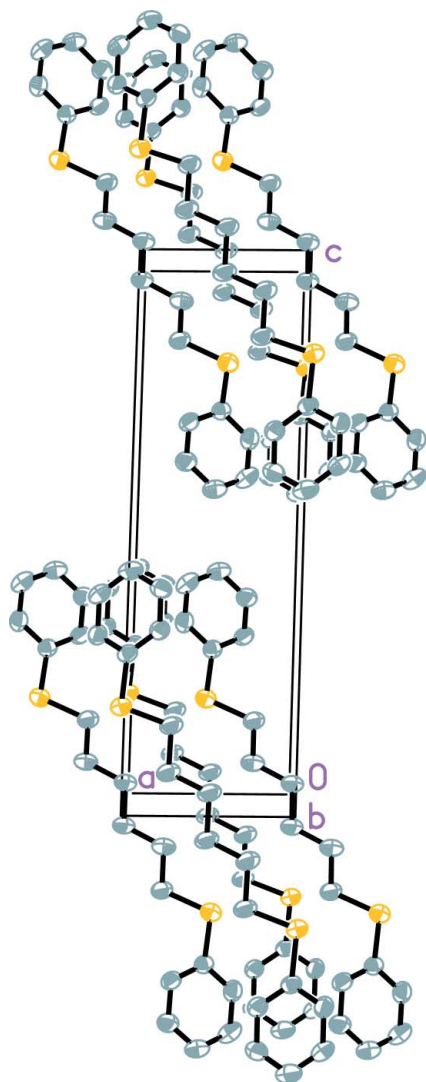


Figure 2

The crystal packing of L^6 , viewed along the b axis. H atoms have been omitted.

$-x, 1 - y, -z]$ and $C29-C29^{ii}$ [symmetry code: (ii) $1 - x, -y, 2 - z]$. The torsion angles in the aliphatic segment of L^6 are all *trans*, indicating that the molecules are in the fully extended conformation (Table 1). The dihedral angles between the aromatic groups and the corresponding S-(CH₂)₆-S planes are 19.4 (2)° for both molecules. The phenyl groups of neighbouring molecules form a 61.5 (1)° dihedral angle. The bond distances and angles in L^6 are in the normal range (Table 1). The crystal packing of L^6 is depicted in Fig. 2. There are no significant π -stacking interactions in the crystal structure.

Experimental

The title compound, L^6 , was synthesized according to a published procedure (Hartley *et al.*, 1979). L^6 was found to be pure from NMR in acetone-*d*₆ (¹H). The compound was obtained as a crystalline powder from which platelet-shaped crystals were gathered. Several crystals were examined, but only one was of barely suitable quality for X-ray analysis (yield: 87%). Analysis found: C 71.43, H 7.32%; calculated for C₁₈H₂₂S₂: C 71.47, H 7.33%. ¹H NMR (acetone-*d*₆): δ

1.46 [*qt*, 4H, -S-CH₂-CH₂-(CH₂)₂-CH₂-CH₂-S-], 1.63 [*qt*, 4H, -S-CH₂-(CH₂)-CH₂-CH₂-(CH₂)-CH₂-S-], 2.95 [*t*, 4H, -S-(CH₂)-CH₂-CH₂-CH₂-CH₂-(CH₂)-S-], 7.14–7.34 (*m*, 10H, C₆H₅-).

Crystal data

C₁₈H₂₂S₂
 $M_r = 302.48$
 Triclinic, $P\bar{1}$
 $a = 5.627$ (2) Å
 $b = 7.862$ (3) Å
 $c = 18.486$ (6) Å
 $\alpha = 94.55$ (3)°
 $\beta = 91.36$ (3)°
 $\gamma = 90.46$ (3)°
 $V = 815.0$ (5) Å³

$Z = 2$
 $D_x = 1.233$ Mg m⁻³
 Cu $K\alpha$ radiation
 Cell parameters from 25 reflections
 $\theta = 20.0$ – 25.0 °
 $\mu = 2.84$ mm⁻¹
 $T = 298$ (2) K
 Platelet, colorless
 $0.23 \times 0.15 \times 0.02$ mm

Data collection

Enraf–Nonius CAD-4 diffractometer
 ω scans
 Absorption correction: Gaussian
 $T_{\min} = 0.560$, $T_{\max} = 0.940$
 25176 measured reflections
 3086 independent reflections
 1270 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.059$
 $\theta_{\text{max}} = 69.8$ °
 $h = -6 \rightarrow 6$
 $k = -9 \rightarrow 9$
 $l = -22 \rightarrow 22$
 5 standard reflections
 frequency: 60 min
 intensity decay: none

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.036$
 $wR(F^2) = 0.068$
 $S = 0.66$
 3086 reflections
 181 parameters

H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.0003P)^2]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.16$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.25$ e Å⁻³

Table 1

Selected geometric parameters (Å, °).

S1–C11	1.764 (3)	S2–C27	1.798 (3)
S1–C17	1.802 (3)	C19–C19 ⁱ	1.524 (6)
S2–C21	1.766 (3)	C29–C29 ⁱⁱ	1.512 (6)
C11–S1–C17	104.78 (17)	C21–S2–C27	105.51 (17)
S1–C17–C18–C19	−179.4 (2)	S2–C27–C28–C29	179.7 (2)
C17–C18–C19–C19 ⁱ	179.9 (4)	C27–C28–C29–C29 ⁱⁱ	−179.7 (4)

Symmetry codes: (i) $-x, -y + 1, -z$; (ii) $-x + 1, -y, -z + 2$.

The poor quality of the crystal and the fact that only 41% of the measured reflections have $I > 2\sigma(I)$ account for the low S value. H atoms were positioned geometrically (C–H = 0.93–0.97 Å) and were included in the refinement in the riding-model approximation; their displacement parameters were set at $1.2U_{\text{eq}}$ of the parent C atoms. A final verification of possible voids was performed using the VOID routine of the *PLATON* program (Spek, 2003).

Data collection: *CAD-4 Software* (Enraf–Nonius, 1989); cell refinement: *CAD-4 Software*; data reduction: local Program; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1997); software used to prepare material for publication: *UdMX* (Maris, 2004).

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